Ramcar Australia & New Zealand

Chemwatch Hazard Alert Code: 4

Issue Date: 16/11/2021

Chemwatch: 47-9809

Version No: 8.1 Print Date: 19/11/2021 Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements L.GHS.AUS.EN

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier

Product name	Ramcar Sulfuric Acid (10%-51%)
Chemical Name	Not Applicable
Synonyms	Not Available
Proper shipping name	SULPHURIC ACID with not more than 51% acid or BATTERY FLUID, ACID
Chemical formula	Not Applicable
Other means of identification	Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Electroplating, fertiliser manufacture, laboratory reagent, pickling and anodising metals.

Details of the supplier of the safety data sheet

Registered company name	Ramcar Australia & New Zealand
Address	Unit A, 1 Reconciliation Rise Pemulwuy NSW 2145 Australia
Telephone	+61 2 9840 2800
Fax	Not Available
Website	www.independentbatterydistributors.com.au; www.lionbatteries.com.au
Email	whsercw@ramcar.com.au

Emergency telephone number

Association / Organisation	CHEMWATCH EMERGENCY RESPONSE	
Emergency telephone numbers	+61 2 9186 1132	
Other emergency telephone numbers	+61 1800 951 288	

Once connected and if the message is not in your prefered language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture	
Poisons Schedule	S6
Classification ^[1]	Skin Corrosion/Irritation Category 1A, Serious Eye Damage/Eye Irritation Category 1, Acute Toxicity (Inhalation) Category 3
Legend:	1. Classified by Chernwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

Label elements

|--|

Signal word Danger

Hazard statement(s)

H314	Causes severe skin burns and eye damage.
H331	Toxic if inhaled.

Precautionary statement(s) Prevention

P260	Do not breathe mist/vapours/spray.
P264	Wash all exposed external body areas thoroughly after handling.
P271	Use only outdoors or in a well-ventilated area.
P280	Wear protective gloves, protective clothing, eye protection and face protection.

Precautionary statement(s) Response

P301+P330+P331	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.	
P303+P361+P353	IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].	
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P310	Immediately call a POISON CENTER/doctor/physician/first aider.	
P363	Wash contaminated clothing before reuse.	
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.	

Precautionary statement(s) Storage

P403+P233	Store in a well-ventilated place. Keep container tightly closed.
P405	Store locked up.

Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
7664-93-9	15-51	sulfuric acid
7732-18-5	49-85	water
Legend:	 Classified by Chernwatch; 2. Classification drawn from HCIS; 3. Cla Classification drawn from C&L * EU IOELVs available 	ssification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4.

SECTION 4 First aid measures

Eye Contact	 If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay. Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema. Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs). As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested. Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSC13719)
Ingestion	 For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Transport to hospital or doctor without delay.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

For acute or short term repeated exposures to strong acids:

- Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling
- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise. ۲
- Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues. INGESTION:
- Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.

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Ramcar Sulfuric Acid (10%-51%)

- DO NOT attempt to neutralise the acid since exothermic reaction may extend the corrosive injury.
- Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
- Charcoal has no place in acid management.
- Some authors suggest the use of lavage within 1 hour of ingestion.

SKIN:

- Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.
- Deep second-degree burns may benefit from topical silver sulfadiazine.

EYE:

- Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjuctival cul-de-sacs. Irrigation should last at least 20-30 minutes. DO NOT use neutralising agents or any other additives. Several litres of saline are required.
- Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury.
- Steroid eye drops should only be administered with the approval of a consulting ophthalmologist).

[Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 Firefighting measures

Extinguishing media

- Water spray or fog.
- Foam.
- Dry chemical powder.BCF (where regulations permit).
- Carbon dioxide.

Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.
Advice for firefighters	

Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Use fire fighting procedures suitable for surrounding area. Do not approach containers suspected to be hot. Cool fire exposed containers with water spray from a protected location. If safe to do so, remove containers from path of fire. Equipment should be thoroughly decontaminated after use.
Fire/Explosion Hazard	 Non combustible. Not considered to be a significant fire risk. Acids may react with metals to produce hydrogen, a highly flammable and explosive gas. Heating may cause expansion or decomposition leading to violent rupture of containers. May emit corrosive, poisonous fumes. May emit acrid smoke. Decomposition may produce toxic fumes of: sulfur oxides (SOx)
HAZCHEM	2R

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material. Check regularly for spills and leaks. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up. Place in a suitable, labelled container for waste disposal. 						
	Chemical Class:acidic compounds For release onto land: recommend SORBENT TYPE RANK APPLI LAND SPILL - SMALL	ded so	OR CC	DLLECTION	LIMITATIONS		
Major Spills	foamed glass - pillows expanded mineral - particulate	1	throw shovel	pitchfork shovel	R, P, DGC, RT R, I, W, P, DGC		
	foamed glass - particulate	2	shovel	shovel	R, W, P, DGC		
	LAND SPILL - MEDIUM						
	expanded mineral -particulate	1	blower	skiploader	R, I, W, P, DGC		
	foamed glass- particulate		blower	skiploader	R, W, P, DGC		

foamed glass - particulate 3 throw skiploade	R, W, P, DGC
Legend	
DGC: Not effective where ground cover is dense	
R; Not reusable	
I: Not incinerable	
P: Effectiveness reduced when rainy	
RT:Not effective where terrain is rugged	
SS: Not for use within environmentally sensitive sites	
W: Effectiveness reduced when windy	
Reference: Sorbents for Liquid Hazardous Substance Cleanup	and Control;
R.W Melvold et al: Pollution Technology Review No. 150: Nove	s Data Corporation 1988
Clear area of personnel and move upwind.	
Alert Fire Brigade and tell them location and nature of haza	ırd.
Wear full body protective clothing with breathing apparatus	
Prevent, by any means available, spillage from entering dra	ains or water course.
Stop leak if safe to do so.	
Contain spill with sand, earth or vermiculite.	
Collect recoverable product into labelled containers for recoverable	/cling.
Neutralise/decontaminate residue (see Section 13 for special	ific agent).
Collect solid residues and seal in labelled drums for dispos	al.
Wash area and prevent runoff into drains.	
After clean up operations, decontaminate and launder all p	rotective clothing and equipment before storing and re-using.
If contamination of drains or waterways occurs, advise emerged	ergency services.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material. Avoid smoking, naked lights or ignition sources. Avoid contact with incompatible materials. When handling, DO NOT eat, drink or smoke. Keep containers securely sealed when not in use. Avoid physical damage to containers. Always wash hands with soap and water after handling. Work clothes should be laundered separately. Launder contaminated clothing before re-use. Use good occupational work practice. Observe manufacturer's storage and handling recommendations contained within this SDS. Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.
Other information	 Store in original containers. Keep containers securely sealed. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this SDS.

Conditions for safe storage, including any incompatibilities

conditions for sale storage, in	cluding any incompatibilities
Suitable container	 DO NOT use aluminium or galvanised containers Check regularly for spills and leaks Lined metal can, lined metal pail/ can. Plastic pail. Polyliner drum. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks. For low viscosity materials Drums and jerricans must be of the non-removable head type. Where a can is to be used as an inner package, the can must have a screwed enclosure. For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.): Removable head packaging; Cans with friction closures and low pressure tubes and cartridges may be used. - Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.
Storage incompatibility	 Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0. Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts - neutralisation can generate dangerously large amounts of heat in small spaces. The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat. The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting "bumping" can spatter the acid. Inorganic acids react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas. Inorganic acids react with cyanide compounds to release gaseous hydrogen cyanide. Inorganic acids generate flammable and/or toxic gases in contact with dithiocarbamates, isocyanates, mercaptans, nitrides, nitriles, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H2S and SO3), dithionites (SO2), and even carbonates.

Acids often catalyse (increase the rate of) chemical reactions.
Reacts with mild steel, galvanised steel / zinc producing hydrogen gas which may form an explosive mixture with air.
Avoid any contamination of this material as it is very reactive and any contamination is potentially hazardous
Sulfuric acid :
is a strong oxidiser
reacts with water or steam
reacts violently with many substances including reducing agents, combustible materials, organic substances, alkalis, ammonium tetraperoxochromate, aniline, 1,2-ethanediamine, ethanolamine, isoprene, mesityl oxide, endonorbanecarboxylic acid ethyl ester, perchlorates, sodium carbonate, zinc chlorate
reacts, possibly causing ignition or explosion, with many substances, including non-oxidising mineral acids, organic acids, bases, reducing agents, acetic anhydride, acetone cyanohydrin, acetonitrile, acrolein, acrylates, acrylonitrile, alcohols, aldehydes, alkylene oxides, allyl alcohol, allyl chloride, substituted allyls, 2-aminoethanol, ammonium hydroxide, bromine pentafluoride, n-butyraldehyde, caprolactam solution, carbides, caesium acetylene carbide, chlorine trifluoride, chlorates, chlorosulfonic acid, cresols, cuprous nitride, diisobutylene, ethylene cyanohydrin, ethylene diamine, ethylene glycol, ethylenemine, fulminates, glycols, hydrochloric acid, iodine heptafluoride, iron, isocyanates, ketones, lithium silicide, mercuric nitride, 2-methyllactonitrile, powdered metals, nitric acid, p-nitrotoluene, pentasilver trihydroxydiaminophosphate, perchloric acid, phenols, phosphorus, picrates, potassium chlorate, potassium permanganate, beta-propiolactone, propylene oxide, pyridine, rubidium acetylene, silver permanganate, sodium, sodium chlorate, sodium hydroxide, styrene monomer, zinc phosphide
increases the explosive sensitivity of nitromethane
incompatible with 2-amino-5-nitrothiazole, 2-aminothiazole, ammonia, aliphatic amines, alkanolamines, amides, organic anhydrides, isocyanate, vinyl acetate, alkylene oxides, epichlorohydrin
attacks some plastics, rubber and coatings
reacts with metals to produce flammable hydrogen gas

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA							
Source	Ingredient	Material name	TWA	STEL	Pe	ak	Notes
Australia Exposure Standards	sulfuric acid	Sulphuric acid	1 mg/m3	3 mg/m3	No	ot Available	Not Available
Emergency Limits							
Ingredient	TEEL-1		TEEL-2			TEEL-3	
sulfuric acid	Not Available		Not Available			Not Available	
Ingredient	Original IDLH			Revised IDLH			
sulfuric acid	15 mg/m3			Not Available			
water	Not Available			Not Available			

MATERIAL DATA

posure controis			
	Engineering controls are used to remove a hazard or place a be highly effective in protecting workers and will typically be The basic types of engineering controls are: Process controls which involve changing the way a job activi Enclosure and/or isolation of emission source which keeps a "adds" and "removes" air in the work environment. Ventilation ventilation system must match the particular process and chh Employers may need to use multiple types of controls to pre- Local exhaust ventilation usually required. If risk of overexpo protection. Supplied-air type respirator may be required in sp An approved self contained breathing apparatus (SCBA) ma Provide adequate ventilation in warehouse or closed storage velocities which, in turn, determine the "capture velocities" of	independent of worker interactions to provide this high level ty or process is done to reduce the risk. selected hazard "physically" away from the worker and ver n can remove or dilute an air contaminant if designed prope emical or contaminant in use. vent employee overexposure. sure exists, wear approved respirator. Correct fit is essentia vecial circumstances. Correct fit is essential to ensure adequ y be required in some situations. area. Air contaminants generated in the workplace posses	of protection. Itilation that strategically rly. The design of a I to obtain adequate uate protection. s varying "escape"
	Type of Contaminant:		Air Speed:
	solvent, vapours, degreasing etc., evaporating from tank (i	0.25-0.5 m/s (50-100 f/min.)	
Appropriate engineering controls	aerosols, fumes from pouring operations, intermittent contr drift, plating acid fumes, pickling (released at low velocity in	0.5-1 m/s (100-200 f/min.)	
	direct spray, spray painting in shallow booths, drum filling, generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)	
	grinding, abrasive blasting, tumbling, high speed wheel ge very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)	
	Within each range the appropriate value depends on:		
	Lower end of the range	Upper end of the range	
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents	
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity	
	3: Intermittent, low production.	3: High production, heavy use	
	4: Large hood or large air mass in motion 4: Small hood-local control only		
	Simple theory shows that air velocity falls rapidly with distance with the square of distance from the extraction point (in simp accordingly, after reference to distance from the contamination 1-2 m/s (200-400 f/min) for extraction of solvents generated	le cases). Therefore the air speed at the extraction point shing source. The air velocity at the extraction fan, for example	ould be adjusted, , should be a minimum

	producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.
Personal protection	
Eye and face protection	 Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material goggles.whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted. Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection. Alternatively a gas mask may replace splash goggles and face shields. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]
Skin protection	See Hand protection below
Hands/feet protection	 Elbow length PVC gloves When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots. The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended. Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: frequency and duration of contact, chemical resistance of glove material, glove thickness and dexterity Select gloves tosted to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent). When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to D 374, AS/NZS 2161.10.1 or national equivalent) is recommended. When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to D 374, AS/NZS 2161.10.1 or national equivalent) is recommended. Some glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term use. Contaminated gloves should be
Body protection	See Other protection below
Other protection	 Overalls. PVC Apron. PVC protective suit may be required if exposure severe. Eyewash unit.

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index". The effect(s) of the following substance(s) are taken into account in the *computer*-

generated selection: Ramcar Sulfuric Acid (10%-51%)

Material	СРІ
NEOPRENE	А
BUTYL	С
NATURAL RUBBER	С

Respiratory protection

Type E-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum	Half-Face	Full-Face	Powered Air
Protection Factor	Respirator	Respirator	Respirator
up to 10 x ES	E-AUS P2	-	E-PAPR-AUS / Class 1 P2

NATURAL+NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
PE	С
PVA	С
PVC	С
SARANEX-23	С
VITON	С

up to 50 x ES	-	E-AUS / Class 1 P2	-
up to 100 x ES	-	E-2 P2	E-PAPR-2 P2 ^

^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

 $\ensuremath{\text{NOTE}}$: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Colourless to brown alkaline liquid with slight odour; mixes with water.		
Physical state	Liquid	Relative density (Water = 1)	1.2-1.4
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	<1	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Negligible	Gas group	Not Available
Solubility in water	Miscible	pH as a solution (%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Applicable

SECTION 10 Stability and reactivity

Reactivity	See section 7	
Chemical stability	Contact with alkaline material liberates heat	
Possibility of hazardous reactions	section 7	
Conditions to avoid	See section 7	
Incompatible materials	See section 7	
Hazardous decomposition products	See section 5	

SECTION 11 Toxicological information

Information on toxicological effects

Inhaled	Acidic corrosives produce respiratory tract irritation with coughing, choking and mucous membrane damage. Symptoms of exposure may include dizziness, headache, nausea and weakness. In more severe exposures, pulmonary oedema may be evident either immediately or after a latent period of 5-72 hours. Symptoms of pulmonary oedema include a tightness in the chest, dyspnoea, frothy sputum and cyanosis. Examination may reveal hypotension, a weak and rapid pulse and moist rates. Death, due to anoxia, may occur several hours after onset of the pulmonary oedema. Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may produce toxic effects; these may be fatal.
	Evidence shows, or practical experience predicts, that the material produces irritation of the respiratory system, in a substantial number of individuals, following inhalation. In contrast to most organs, the lung is able to respond to a chemical insult by first removing or neutralising the irritant and then repairing the damage. The repair process, which initially evolved to protect mammalian lungs from foreign matter and antigens, may however, produce further lung damage resulting in the impairment of gas exchange, the primary function of the lungs. Respiratory tract irritation often results in an inflammatory response involving the recruitment and activation of many cell types, mainly derived from the vascular

	system. Exposure to high concentrations causes bronchitis and is characterised by th	ne onset of haem	orrhagic pulmonary oedema.	
Ingestion	Ingestion of acidic corrosives may produce circumoral burns with a distinct discolouration of the mucous membranes of the mouth, throat and oesophagus. Immediate pain and difficulties in swallowing and speaking may also be evident. Oedema of the epiglottis may produce respiratory distress and possibly, asphyxia. Nausea, vomiting, diarrhoea and a pronounced thirst may occur. More severe exposures may produce a vomitus			
Skin Contact	Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue. Open cuts, abraded or irritated skin should not be exposed to this material			
Eye	When applied to the eye(s) of animals, the material produces severe ocular lesions which are present twenty-four hours or more after instillation. Direct eye contact with acid corrosives may produce pain, lachrymation, photophobia and burns. Mild burns of the epithelia generally recover rapidly and completely. Severe burns produce long-lasting and possible irreversible damage. The appearance of the burn may not be apparent for several weeks after the initial contact. The cornea may ultimately become deeply vascularised and opaque resulting in blindness.			
Chronic	Repeated or prolonged exposure to acids may result in the erosion of teeth, (rarely) of the jaw. Bronchial irritation, with cough, and frequent attacks of bro also occur. Chronic exposures may result in dermatitis and/or conjunctivitis. The impact of inhaled acidic agents on the respiratory tract depends upon a characteristics, e.g., gas versus aerosol; particle size (small particles can pe are more likely to be removed in the nose and mouth). Given the general lac occupational exposures to acids, it is difficult to identify their principal deposi with a diameter of up to a few micrometers will be deposited in both the uppe cause dental erosion, and they produce acute effects in the lungs (symptoms particular risk for pulmonary effects. Limited evidence suggests that repeated or long-term occupational exposure biochemical systems.	number of interre netrate deeper in k of information c tion site within th er and lower airwa s and changes in	a may ensue. Gastrointestinal disturbances may lated factors. These include physicochemical to the lung); water solubility (more soluble agents on the particle size of aerosols involved in e respiratory tract. Acid mists containing particles ays. They are irritating to mucous epithelia, they pulmonary function). AsthmatIcs appear to be at	
	TOXICITY	RITATION		
Ramcar Sulfuric Acid (10%-51%)		lot Available		
sulfuric acid	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1]	rritation ye (rabbit): 1.38 i ye (rabbit): 5 mg/	-	
sulfuric acid	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E	ye (rabbit): 1.38 i ye (rabbit): 5 mg/	-	
sulfuric acid water	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF	ye (rabbit): 1.38 i ye (rabbit): 5 mg/ RRITATION	-	
	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF	ye (rabbit): 1.38 i ye (rabbit): 5 mg/	-	
	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF	ye (rabbit): 1.38 i ye (rabbit): 5 mg, RRITATION lot Available y 2.* Value obtair	30sec SEVERE	
water	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF Oral(Rat) LD50; >90000 mg/kg ^[2] N 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity	ye (rabbit): 1.38 i ye (rabbit): 5 mg/ RRITATION lot Available y 2.* Value obtain Substances e to the material in occur following ecceding respirato nented exposure y on methacholin e criteria for diag ration of and dura exposure due to e disorder is char	30sec SEVERE	
water Legend:	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF Oral(Rat) LD50; >90000 mg/kg ^[2] N 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity specified data extracted from RTECS - Register of Toxic Effect of chemical S Occupational exposures to strong inorganic acid mists of sulfuric acid: Asthma-like symptoms may continue for months or even years after exposure condition known as reactive airways dysfunction syndrome (RADS) which car compound. Key criteria for the diagnosis of RADS include the absence of pre onset of persistent asthma-like symptoms within minutes to hours of a docum spirometry, with the presence of moderate to severe bronchial hyperreactivity lymphocytic inflammation, without eosinophilia, have also been included in the irritating inhalation is an infrequent disorder with rates related to the concent Industrial bronchitis, on the other hand, is a disorder that occurs as result of particulate in nature) and is completely reversible after exposure ceases. The production.	ye (rabbit): 1.38 i ye (rabbit): 5 mg/ RRITATION lot Available y 2.* Value obtain Substances e to the material in occur following ecceding respirato nented exposure y on methacholin e criteria for diag ration of and dura exposure due to e disorder is char	30sec SEVERE	
water Legend: SULFURIC ACID	Inhalation(Mouse) LC50; 0.85 mg/4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF Oral(Rat) LD50; >90000 mg/kg ^[2] N 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity specified data extracted from RTECS - Register of Toxic Effect of chemical S Occupational exposures to strong inorganic acid mists of sulfuric acid: Asthma-like symptoms may continue for months or even years after exposur condition known as reactive airways dysfunction syndrome (RADS) which ca compound. Key criteria for the diagnosis of RADS include the absence of pre onset of persistent asthma-like symptoms within minutes to hours of a docum spirometry, with the presence of moderate to severe bronchial hyperreactivity lymphocytic inflammation, without eosinophilia, have also been included in the irritating inhalation is an infrequent disorder with rates related to the concent Industrial bronchitis, on the other hand, is a disorder that occurs as result of particulate in nature) and is completely reversible after exposure ceases. The production. WARNING: For inhalation exposure ONLY: This substance has been classified not be seen characteria.	ye (rabbit): 1.38 i ye (rabbit): 5 mg/ RRITATION lot Available y 2.* Value obtain Substances e to the material in occur following ecceding respirato nented exposure y on methacholin e criteria for diag ration of and dura exposure due to e disorder is char	30sec SEVERE	
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water <i>Legend:</i> SULFURIC ACID WATER Acute Toxicity	Inhalation(Mouse) LC50; 0.85 mg/l4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF Oral(Rat) LD50; >90000 mg/kg ^[2] N 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity specified data extracted from RTECS - Register of Toxic Effect of chemical S Occupational exposures to strong inorganic acid mists of sulfuric acid: Asthma-like symptoms may continue for months or even years after exposur condition known as reactive airways dysfunction syndrome (RADS) which ca compound. Key criteria for the diagnosis of RADS include the absence of prosest of persistent asthma-like symptoms within minutes to hours of a docum spirometry, with the presence of moderate to severe bronchial hyperreadtint tirritating inhalation is an infrequent disorder with rates related to the concent Industrial bronchitis, on the other hand, is a disorder that occurs as result of particulate in nature) and is completely reversible after exposure ceases. The production. WARNING: For inhalation exposure ONLY: This substance has been classifiered and the toxicological data identified in literature search.	ye (rabbit): 1.38 i ye (rabbit): 5 mg/ RRITATION lot Available y 2.* Value obtain Substances e to the material in occur following ecceding respirato nented exposure y on methacholing ecretiral for dilay exposure due to e disorder is char ied by the IARC a	30sec SEVERE and from manufacturer's SDS. Unless otherwise ceases. This may be due to a non-allergenic exposure to high levels of highly irritating ry disease, in a non-atopic individual, with abrupt to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal mosis of RADS. RADS (or asthma) following an tion of exposure to the irritating substance. high concentrations of irritating substance (often acterised by dyspnea, cough and mucus is Group 1: CARCINOGENIC TO HUMANS	
water <i>Legend:</i> SULFURIC ACID WATER Acute Toxicity Skin Irritation/Corrosion	Inhalation(Mouse) LC50; 0.85 mg/4h ^[1] E Oral(Rat) LD50; >300 mg/kg ^[1] E TOXICITY IF Oral(Rat) LD50; >90000 mg/kg ^[2] N 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity specified data extracted from RTECS - Register of Toxic Effect of chemical S Occupational exposures to strong inorganic acid mists of sulfuric acid: Asthma-like symptoms may continue for months or even years after exposure condition known as reactive airways dysfunction syndrome (RADS) which ca compound. Key criteria for the diagnosis of RADS include the absence of preonset of persistent asthma-like symptoms within minutes to hours of a docum spirometry, with the presence of moderate to severe bronchial hyperreactivity lymphocytic inflammation, without eosinophilia, have also been included in the irritating inhalation is an infrequent disorder with rates related to the concent Industrial bronchitis, on the other hand, is a disorder that occurs as result of particulate in nature) and is completely reversible after exposure ceases. The production. WARNING: For inhalation exposure ONLY: This substance has been classified in a cute toxicological data identified in literature search. Image: Stort - Sime sto	ye (rabbit): 1.38 i ye (rabbit): 5 mg, RRITATION lot Available y 2.* Value obtain Substances e to the material an occur following ecceding respirato nented exposure y on methacholin e criteria for diag ration of and dure exposure due to e disorder is char ed by the IARC a arcinogenicity teproductivity	30sec SEVERE and from manufacturer's SDS. Unless otherwise exposure to high levels of highly irritating ry disease, in a non-atopic individual, with abrupt to the irritant. A reversible airflow pattern, on e challenge testing and the lack of minimal nosis of RADS. RADS (or asthma) following an tion of exposure to the irritating substance. ingh concentrations of irritating substance (often acterised by dyspnea, cough and mucus as Group 1: CARCINOGENIC TO HUMANS	

Data available to make classification

SECTION 12 Ecological information

Toxicity

	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
	NOEC(ECx)	Not Available	Crustacea	0.15mg/l	2
sulfuric acid	EC50	72h	Algae or other aquatic plants	2.56mg/l	2
	LC50	96h	Fish	0.75mg/l	2
	EC50	48h	Crustacea	3.05mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
water	Not Available	Not Available	Not Available	Not Available	Not Available
Legend:	V3.12 (QSAR)	1. IUCLID Toxicity Data 2. Europe ECHA Registe - Aquatic Toxicity Data (Estimated) 4. US EPA, Ec Japan) - Bioconcentration Data 7. METI (Japan) -	otox database - Aquatic Toxicity Data 5. ECET		

Prevent, by any means available, spillage from entering drains or water courses. DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil Persistence: Air			
water	LOW LOW			
Discourse lating and and a				
Bioaccumulative potential				
Ingredient	Bioaccumulation			
	No Data available for all ingredients			
Mobility in soil	Mobility in soil			
Ingredient	Mobility			
	No Data available for all ingredients			

SECTION 13 Disposal considerations

	Containers may still present a chemical hazard/ danger when empty.
	Return to supplier for reuse/ recycling if possible.
	Otherwise:
	If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
	Where possible retain label warnings and SDS and observe all notices pertaining to the product.
	Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.
	A Hierarchy of Controls seems to be common - the user should investigate:
	▶ Reduction
	▶ Reuse
	▶ Recycling
	Disposal (if all else fails)
	This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been
Product / Packaging disposal	contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be
	applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be
	appropriate.
	DO NOT allow wash water from cleaning or process equipment to enter drains.
	It may be necessary to collect all wash water for treatment before disposal.
	In all cases disposal to sever may be subject to local laws and regulations and these should be considered first.
	• Where in doubt contact the responsible authority.
	Recycle wherever possible.
	Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
	Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial
	in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
	 Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until
	containers are cleaned and destroyed.

SECTION 14 Transport information

Labels Required



Marine Pollutant

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Ramcar Sulfuric Acid (10%-51%)

HAZCHEM 2R

Land transport (ADG)			
UN number	2796		
UN proper shipping name	SULPHURIC ACID with not more than 51% acid or BATTERY FLUID, ACID		
Transport hazard class(es)	Class 8 Subrisk Not Applicable		
Packing group	I		
Environmental hazard	Not Applicable		
Special precautions for user	Special provisions Not Applicable Limited quantity 1 L		

Air transport (ICAO-IATA / DGR)

UN number	2796			
UN proper shipping name	Sulphuric acid with 51% or less acid; Battery fluid, acid			
Transport hazard class(es)	ICAO/IATA Class ICAO / IATA Subrisk ERG Code	8 Not Applicable 8L		
Packing group				
Environmental hazard	Not Applicable			
	Special provisions Cargo Only Packing Instructions		Not Applicable	
	Cargo Only Maximum Qty / Pack		30 L	
Special precautions for user			851	
	Passenger and Cargo Maximum Qty / Pack		1 L	
	Passenger and Cargo	Limited Quantity Packing Instructions	Y840	
	Passenger and Cargo Limited Maximum Qty / Pack		0.5 L	

Sea transport (IMDG-Code / GGVSee)

UN number	2796		
UN proper shipping name	SULPHURIC ACID with not more than 51% acid or BATTERY FLUID, ACID		
Transport hazard class(es)	IMDG Class 8 IMDG Subrisk Not Applicable		
Packing group	1		
Environmental hazard	Not Applicable		
Special precautions for user	EMS Number Special provisions Limited Quantities		

Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group			
sulfuric acid	Not Available			
water	Not Available			

Transport in bulk in accordance with the ICG Code

Product name	Ship Type
sulfuric acid	Not Available
water	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

sulfuric acid is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -Schedule 6

Australian Inventory of Industrial Chemicals (AIIC)

Chemical Footprint Project - Chemicals of High Concern List

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs - Group 1: Carcinogenic to humans

water is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

National Inventory Status

National Inventory	Status	
Australia - AIIC / Australia Non-Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (sulfuric acid; water)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

SECTION 16 Other information

Revision Date	16/11/2021
Initial Date	26/02/2015

SDS Version Summary

Version	Date of Update	Sections Updated
7.1	09/11/2021	Name
8.1	16/11/2021	Name

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit. IDLH: Immediately Dangerous to Life or Health Concentrations ES: Exposure Standard OSF: Odour Safety Factor NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index AIIC: Australian Inventory of Industrial Chemicals DSL: Domestic Substances List NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas NCI: National Chemical Inventory FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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